

Time allowed: 3 Hours

Max. Marks: 50

NOTE: Attempt five questions in all, including Question No. 1 which is compulsory and selecting two questions from each Part.

x-x-x

I. Attempt any five parts (2 x 5=10)

- Sketch $(11\bar{2})$ plane within a cubic unit cell.
- Differentiate between zig-zag and armchair carbon nanotubes.
- What do you understand by viscoelasticity?
- How do factors such as temperature and stress affect the creep strain?
- Define eutectoid transformation and provide an example of a material system where it occurs.
- Under what conditions we get coarse pearlite and fine pearlite from austenite?

Part-A

- II. (a) What are ionic crystals? Derive an expression for the cohesive energy for an ionic crystal.

(b) Describe a hexagonal close pack structure and calculate its packing fraction.

(5,5)

- III. (a) What are point defects in crystal lattice? How are they caused?

(b) Give a brief account of phenomenon of diffusion. How is diffusion employed for doping in semiconductors?

(3,7)

- IV. (a) Discuss the atomistic basis of elastic behaviour.

(b) A tensile load of 100 N is applied to an aluminium-boron composite of 1 mm^2 cross-sectional area. The volume of the parallel fibres is 30%. What is the stress in the fibres, when the load axis is (i) parallel to the fibres, and (ii) perpendicular to the fibres? (Given, $Y_{\text{Aluminium}}=71 \text{ GN/m}^2$ and $Y_{\text{Boron}}=440 \text{ GN/m}^2$)

(6,4)

Part-B

- V. (a) What do you understand by slip system? Describe the mechanism of slip in a single crystal.

(b) Describe the mechanisms responsible for the ductile to brittle transition in metals

Discuss the factors that influence the ductile to brittle transition temperature. (6,4)

- VI. Consider the binary isomorphous solution of copper and nickel. Draw its phase diagram and discuss the various microstructures developed during its equilibrium and non-equilibrium cooling. (10)

- VII. (a) Find free energy change associated with homogeneous nucleation of a sphere of radius r . Find the critical nucleus radius r^* for homogeneous nucleation.

(b) Briefly cite the differences between pearlite, bainite and spheroidite relative to microstructure and mechanical properties. (6,4)

x-x-x