

Exam.Code:0940
Sub. Code: 6711

2014
B.E. (Mechanical Engineering)
Fourth Semester
MEC-401: Engineering Thermodynamics

Time allowed: 3 Hours

Max. Marks: 50

NOTE: Attempt five questions in all, including Question No. 1 which is compulsory and selecting two questions from each Part. Use of property, steam and gas tables is permitted.

x-x-x

Q-1)

- 1) Draw a generalized jet propulsion system with all critical components.
- 2) What is Holzwarth Explosion Turbine?
- 3) What is the significance of clearance volume in cylinders?
- 4) What is meant by compressor turbine?
- 5) Define Irreversibility in mathematical terms.

(5x2=10)

PART A

Q-2)

- 1) Explain with expressions the decrease of exergy principle.
- 2) Show various components of total flow energy with exergy.

(5,5)

Q-3)

- 1) Determine ideal efficiency of a diesel engine having cylinder bore 250mm, stroke 375mm, clearance volume 1500cc, with fuel cut-off occurring at 5% of stroke.
- 2) Derive and compare air standard efficiencies of Otto and Diesel cycles with PV and TS diagrams

(5,5)

Q-4)

- 1) Explain construction, working and advantages of a Turbo prop engine.
- 2) Calculate decrease in exergy when 25kg of water at 95°C mixed with 35kg of water at 35°C, the pressure being taken as constant and temperature of the surroundings being 15°C.

(5,5)

P.T.O.

(2)

PART B

Q-5)

- 1) What are the conclusions of Henry's Law?
- 2) 0.25kg of an ideal gas at 300kPa, 80°C and 0.07m³ undergoes adiabatic process to a final condition of 300kPa and 0.1m³. During this process 25kJ of work is done on the gas. Find the value of R, C_p and C_v for the gas.

(5,5)

Q-6)

- 1) A mixture of 1kmol of H₂O and 2kmol of O₂ is heated at 4000K at 1atm. Determine the equilibrium composition of this mixture, assuming that only H₂O, OH, O₂ and H₂ are present.
- 2) Differentiate between Dalton's and Amagat's laws for predicting P-V-T behavior of a gas mixture.

(5,5)

Q-7)

Write short notes on any 2 of the following:-

- 1) Relate mole, pressure and volume fractions in ideal gas mixtures.
- 2) Flow work and non-flow work
- 3) Clausius and Kelvin Plank statements

(5,5)