

Time allowed: 3 Hours

Max. Marks: 50

NOTE: Attempt five questions in all, including Question No. 1 which is compulsory and selecting two questions from each Unit.

x-x-x

I. Answer briefly: -

- a) Define pyrolysis.
- b) The grid in electron microscope is made up of _____.
- c) Why is degassing done in solvent before chromatography?
- d) Define limit of resolution in a microscope.
- e) Draw the chemical structure of TMS.
- f) What is a chromatogram?
- g) ESR is an acronym for _____.
- h) Name the two monochromators used in fluorescence spectrophotometer.
- i) FRET is an acronym for _____.
- j) Define Larmor frequency.

(10x1)

UNIT - I

- II. a) You are an environmental biologist and had to report about heavy metal pollution in a lake. Which analytical technique you will use to determine the metal ion concentration in the water, explain the working protocol in detail.
- b) Describe the principle of spectrofluorimetry by highlighting the different phases of fluorescence.
- (2x5)
- III. a) Define chemical shift. A compound Y shows a chemical shift of 7.46 ppm using a 90 MHz NMR. Calculate the chemical shift of Y in Hz. What will the chemical shift of Y using a 300 MHz NMR (both in Hz and ppm)?
- b) Explain the n+1 rule using an example. Discuss the reason for J coupling/ spin-spin splitting.
- (2x5)

P.T.O.

(2)

- IV. a) What is Bragg's Law of diffraction and its significance in crystal structure determination? (4)
b) Explain spin-spin and spin-lattice relaxation in MRI. (3)
c) In IR spectroscopy the spectrum ranges from $600\text{-}4000\text{cm}^{-1}$. What is the significance of these two numbers have based on electromagnetic spectrum and how to you get to these numbers? (3)

UNIT - II

- V. a) Differentiate between the working of Scanning tunneling microscope and Atomic force microscope. (5)
b) Discuss the instrumentation details of a mass-spectrometer with the help of a diagram. (5)
- VI. a) Name the two types of detectors in Gas chromatography. Elaborate on each type. (4)
b) Write a descriptive note GM counter and scintillation counters. (6)
- VII. a) How can isotopes be identified using a Mass spectroscopy, Explain using suitable examples. (5)
b) Mention various applications of radioisotopes in biomedical research. (5)

X-X-X