12/6/20 E)

Exam.Code:0910 Sub. Code: 6716

2062

B.E. (Biotechnology) Sixth Semester BIO-613: Bioreactor Design and Operation

Time allowed: 3 Hours

Max. Marks: 50

NOTE: Attempt <u>five</u> questions in all, including Question No. I which is compulsory and selecting two questions from each Section.

x-x-x

1. Write briefly:

 $(1 \times 10 = 10)$

- a) How does theoretical yield differ from observed yield?
- b) What is the use of bourdon tube?
- c) What is riser and downcomer?
- d) What are the components of a control loop?
- e) Why fed batch operation is a better choice for Bakers' yeast production?
- f) Give example of model organism.
- g) Give a graphical presentation of variation of temperature with time for batch sterilization.
- h) How can you relate dilution rate with Monod equation?
- i) For immobilized cells what will be the expression of dilution rate?
- j) What are the materials used in membrane bioreactor?

SECTION-A

- 2. a) How microorganism type and culture characteristics are important for bioreactor design and operations?
 - b) Derive mass balance equation of growth limiting substrate in a batch bioreaction.

(5, 5)

- 3. a) What is the importance of membrane bioreactors? What are the challenges of this type of bioreactor?
 - b) What is the use of trickling bed filter in waste water treatment?

(6, 4)

- 4. i) Saccharomyces cerevisiae is used to convert glucose to ethanol in batch fermenter under anaerobic conditions. The yield of biomass from substrate is 0.06 gg⁻¹; Ypx is 7.7 gg⁻¹. The maintenance coefficient is 2.2 gg⁻¹h⁻¹; the specific rate of product formation due to maintenance is 1.1 h⁻¹. The maximum specific growth rate of S. cerevisiae is approaximately 0.3h⁻¹. 5 g bacteria are inoculated into 50 liters of medium containing 12 gl⁻¹. Determine batch culture time required to:
- a) produce 10 g biomass;
- b) achieve 90% substrate conversion
- c) produce 100 g ethanol.
- ii) How can we determine RTD using pulse input and step input method?

SECTION-B

- 5. a) What are the automatic control systems available? Graphically explain the difference with brief discussion.
 - b) Describe the measurement technique of any three process variable.

(5, 5)

- 6. a) Justify how production kinetics changes with energy metabolism for different products.
 - b) Design continuous sterilization flow modél and graphically represent temperature variation with time.

(5, 5)

- 7. a) Phosphoglucomutase converts glucose 1-phosphate to glucose 6 phosphate during glycogen breakdown. In a 1 litre solution at 25°C, the reaction is started with 0.04 gmol glucose 1-phosphate. The reaction proceed to equilibrium at which the concentration of glucose 1 phosphate is 0.002 M and the concentration of glucose 6 phosphate is 0.038 M.
 - i) Calculate the equilibrium constant
 - ii) What is the theoretical yield?
 - iii) What is the yield based on the reactant supplied?
 - b) Derive the expression of L (length) in a continuous plug flow tubular reactor.

(5, 5)