

2072

B.E. (Biotechnology) Second Semester
ASP-X03: Physics of Materials
(Common with IT,CSE and MEC)

Time allowed: 3 Hours

Max. Marks: 50

NOTE: Attempt five questions in all, including Question No. 1 which is compulsory and selecting two questions from each Part.

x-x-x

- 1 (a) Offer an explanation as to why covalently bonded materials are generally less dense than ionically or metallically bonded ones. (2)
- (b) What is the difference between a lattice and a crystal? (2)
- (c) Rank the magnitudes of the diffusion coefficients from greatest to least for the following systems: (i) N in Fe at 700C (ii) Cr in Fe at 700C (iii) N in Fe at 900C (iv) Cr in Fe at 900C. Also justify this ranking. (2)
- (d) Explain the difference between resolved shear stress and critical resolved shear stress. (2)
- (e) Cite two major differences between martensitic and pearlitic transformations. (2)

Part A

- 2 (a) Niobium has an atomic radius of 0.1430nm and a density of 8.57g/cm³. Determine whether it has an FCC or BCC crystal structure. (4)
- (b) Draw an orthorhombic unit cell, and within that cell a [2 $\bar{1}$ 1] direction. (3)
- (c) Derive planar density expressions for FCC (100) and FCC (111) planes in terms of the atomic radius R. (3)
- 3 (a) What are dislocations? Differentiate between edge and screw dislocations. (4)
- (b) For which set of crystallographic planes will a first-order diffraction peak occur at a diffraction angle of 44.53° for FCC nickel when a monochromatic radiation having a wavelength of 0.1542 nm is used? (3)
- (c) For a steel alloy it has been determined that a carburizing heat treatment of 15 h duration will raise the carbon concentration to 0.35 wt% at a point 2.0 mm from the surface. Estimate the time necessary to achieve the same concentration at a 6.0 mm position for an identical steel and at the same carburizing temperature. (3)
- 4 (a) Show that the void in a simple cube can be filled by sphere having radius not larger than 0.732R, where R is the radius of the larger sphere involved in the packing of these spheres. (4)
- (b) Cite the primary differences between elastic, anelastic, and plastic deformation behaviors. (3)
- (c) A steel bar and an aluminium bar are each under a load of 5000N. If the cross-sectional area of the steel bar is 100 mm², what must be the area of aluminium for the same elastic deformation. Given the young's moduli $E_{Al} = 71 \text{ GN/m}^2$, $E_{steel} = 210 \text{ GN/m}^2$. (3)

Part B

- 5 (a) What is a slip system. Do all metals have the same slip system? Why or why not? (3)
- (b) Explain the differences in grain structure for a metal that has been cold worked and one that has been cold worked and then recrystallized. (3)
- (c) Distinguish between (i) stable and unstable crack (ii) intergranular and transgranular fracture. (4)
- 6 (a) With reference to the Fatigue failure, What information does an SN plot convey? (4)
- (b) Discuss the typical creep curve for a material subjected to a constant load creep test, while maintaining the temperature constant. (4)
- (c) Briefly explain why a proeutectoid phase (ferrite or cementite) forms along austenite grain boundaries. (2)
- 7 (a) Discuss the development of microstructure in Pb-Sn alloy of eutectic composition as it is cooled from a temperature above the eutectic temperature to a temperature below the eutectic temperature. (4)
- (b) Briefly discuss the differences between (i) congruent and incongruent transformations (ii) TTT and CCT transformations. (4)
- (c) What is Temper Embrittlement? (2)

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