Exam.Code:1029 Sub. Code: 7852

#### 1129

# M. Tech. (Material Science and Technology) **First Semester**

MST-102: Material Characterization

Max. Marks: 50

NOTE: Attempt five questions in all, including Question No. I which is compulsory and selecting two auestions from each Hair <sub>Jime allowed:</sub> 3 Hours and selecting two questions from each Unit.

#### x-x-x

- Attempt the following:-I.
  - a) How one can determine particle size with x-ray diffraction?
  - b) In an Auger process, what affects the binding energy of an ejected electron?
  - c) Explain the working principal and construction of an penning-gauge
  - $(4x2\frac{1}{2})$ d) What is the role of precursor and career gas in a CVD process?

### UNIT – I

- a) Define the atomic scattering and geometrical structure factor. II.
- b) Calculate structure factor for a fee unit cell. III.
  - c) Explain possible reasons that cause discrepancies between the peak positions of an X-ray spectrum and its standard data.
- a) Describe the epitaxial growth modes in MBE process. IV.
  - b) In an MBE process discuss the role of RHEED in determining the surface quality of the substrate as well as film thickness.
  - a) Explain the working principal and applications of the thermogravimetric analysis. V.
    - b) List similarities and differences between TGA and DSC.
    - c) Sketch a DSC curve that shows a melting peak with a low and a high heating rate.

## UNIT - II

- a) How elemental identification is performed using the X-ray fluorescence VI. Spectroscopy?
  - b) Impact of molecular symmetry and selection rules on Raman spectra. (5+5)P.T.O.

VII.	Write short notes on any two of the following.	
	a) FTIR spectroscopy	
	b) Particle size determination methods	
	c) Luminescence spectroscopy	(2x5)
VIII.	a) Describe the x-ray fluorescence processes and related selection rules.	
	b) Discuss its application in trace element analysis.	(5,3,2)

*x-x-x*