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Exam.Code:0905
Sub. Code: 6650

1129
B.E. (Mechanical Engineering)
First Semester
CH-101: Applied Chemistry
(Common with ECE and EEE)

Time allowed: 3 Hours

Max. Marks: 50

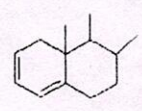
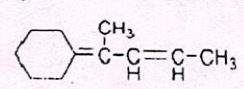
NOTE: Attempt five questions in all, including Question No. 1 which is compulsory and selecting two questions from each Unit.

x-x-x

- I. Answer the following:-
 - a) Explain the factors that can affect the crystallinity of polymers.
 - b) State the disadvantages of crystal field theory
 - c) What is differential aeration corrosion
 - d) Write the difference between auxochrome and chromophore with examples
 - e) Give two definitions of second law of thermodynamics (5x2)

UNIT - I

- II. a) The molar heat capacities at constant pressure of H₂ (g), Cl₂(g) and HCl (g) are 29.3,34.7 and 28.9 JK⁻¹ respectively. If the heat of formation of HCl (g) at constant pressure at 293 K is -91.2 KJ, what will be its heat of formation at 315 K?
- b) Calculate the enthalpy of formation of methane given that the enthalpy of combustion of methane, graphite and hydrogen are -890.2, -393.4 and 285.7 kj/mol at 298 K.
- c) Derive Gibbs-Helmholtz equation. (3,3,4)
- III. a) What is wacker process ? Explain its stepwise mechanism for the catalytic cycle
- b) Derive Michaelis-Menton's equation for enzyme catalysis. When the reaction rate is of first order ? (2x5)
- IV. a) Discuss the effect of the solvent on π - π* and n-π* transition.
- b) Calculate the number of vibrational degrees of freedom in following compounds:
(i) SO₂ (ii) NH₃ (iii) CH₄
- c) Calculate the λ_{max} for the following compounds



(3,3,4)
P.T.O.

(2)

UNIT - II

- V. a) write the difference between thermoset and thermoplastic
b) Explain the mechanism of Zeigler Natta Polymerization.
c) Explain detailed synthesis, properties and uses of epoxy resins. (3,3,4)
- VI. a) write the difference between pitting and waterline corrosion
b) Discuss the construction and working of methanol-oxygen fuel cell
c) Discuss the prevention measures for the corrosion (3,4,3)
- VII. a) Predict the color and spin-only Magnetic Moment for $[\text{Co}(\text{Cl})_4]^{2-}$ and $[\text{Co}(\text{CN})_6]^{3-}$
b) Briefly explain the crystal field splitting in (i) tetrahedral and (ii) square planar complexes.
c) Calculate the CFSE of the following compounds (4,4,2)
i) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$ ii) $[\text{Ni}(\text{CN})_4]^{2+}$

x-x-x