## 1058

B.E. (Bio-Technology)

Eighth Semester

BIO-804: Modeling and Simulation of Bio-Processes

reallowed: 3 Hours

Max. Marks: 50

 $(1 \times 10 = 10)$ 

 $\mathcal{M}^{\text{Hempt}}_{\text{five}}$  questions in all, including Question No. I which is compulsory and selecting  $\mathcal{M}^{\text{He}}_{\text{five}}$  questions from each Section.

x - x - x

Write briefly:

a) Define respiratory quotient (RQ)?

b) Write down steady state and dynamic mass balance equation?

c) What do you understand by structured and un-structured models?

d) What do you understand by data smoothing?

e) Write down characteristics of batch reactor.

f) Reactor choice for zero order kinetics \_\_\_\_\_ and \_\_\_\_

g) List two advantage and disadvantage of fed batch reactor.

h) What is product yield with respect to substrate?

i) Define metabolic engineering.

j) What is the purpose of metabolic Engineering?

## SECTION-A

Assume that the cell can convert 67% of carbon source to biomass. Hexadecane and glucose are used as carbon sources. Calculate the stoichiometric coefficients of following reactions:

 $C_{16}H_{34} + α_1 O_2 + α_2 NH_3 → β_1 (C_{4.4} H_{7.3} N_{0.86} O_{1.2}) + β_2 CO_2 + β_3 H_2 O_2$ 

 $C_6H_{12}O_6 + \alpha'_1O_2 + \alpha'_2NH_3 \rightarrow \beta'_1(C_{4.4}H_{7.3}N_{0.86}O_{1.2}) + \beta'_2CO_2 + \beta'_3H_2O_2$ 

Using the following model equations predict the time.

$$\frac{dx}{dt} = \mu x \qquad \qquad \mu = \frac{\mu_{\max}S}{K_s + S} \qquad \qquad Y_{\frac{x}{s}} = \frac{\Delta X}{\Delta S}$$
(7, 3)

<sup>1</sup>. a)

b)

b)

In an experiment to measure the effect of a drug on leishmaniasis, infected hamsters were given subcutaneous injections at intervals of 3 days in 4 different doses which resulted in 100% survival. A table of the dosage vs parasite burden is given below. From an exponential fit, the drug concentration needed to reduce the parasite load of the spleen to 50% was determined to be 2.5 mg/kg body weight. Would you agree with this estimate?

	0	15	21	57	110
Drug dose (mg/kg body weight)	0	1.5	2.4	5.7	11.2
Diug dose (mg - g	6	5.1	4.9	1.0	0.3
Parasite burden (×10 <sup>-8</sup> )					

Water is flowing into a well-stirred tank at 150 kg/hr and methanol is being added at 30 kg/hr. The resulting solution is leaving the tank at 120 kg/hr, there are 100 kg of fresh water in the tank at the start of the operation and the rates of input and output remain constant thereafter. Calculate the outlet concentration (mass fraction f methanol) after 1 hr. what would be the steady state methanol mass fraction. (6, 4)

- Aspergillus niger is used to convert glucose to gluconic acid via gluconolactone. If the Aspergillus niger is used to convert glucose to gradient with  $K_1$  and  $K_2$  rate constant, derive the conversion follows the first order consecutive reaction with  $K_1$  and  $K_2$  rate constant, derive the 4. a) conversion follows the first order consecutive reaction of intermediate (gluconolactone) is maximized and also derive the  $t_{max}$  at which the concentration of intermediate (gluconolactone. Provided that the initial concentration of  $t_{max}$  at which the concentration of intermediate (b) and the initial concentration of maximum concentration ( $C_{Lmax}$ ) gluconolactone. Provided that the initial concentration of glucose at zero time G<sub>0</sub>.
  - glucose at zero time  $c_0$ . Lipase is being investigated as an additive to laundry detergent for removal of stains from  $f_{abric}$ . b) The general reaction is ---

Fats  $\rightarrow$  Fatty acid + glycerol

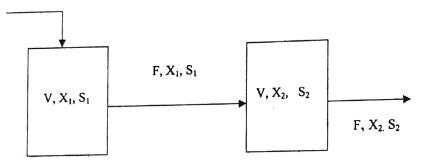
The Michaelis constant for pancreatic lipase is 5 mM. At 60 °C, lipase is subjected to with half life of 8 min. Fat hydrolysis under specific condition which simulates a top - loading washing machine. The initial fat concentration is 45 mM and maximum reaction top - loading washing machine. The initial and does it take for the enzyme to hydrolyse  $80\%_{0f}$  rate of hydrolysis is 0.07 m mol 1<sup>-1</sup> s<sup>-1</sup>. How long does it take for the enzyme to hydrolyse  $80\%_{0f}$ (5, 5)the fat present?

1.

## SECTION-B

Discuss the balance equations for each component as given in two chemostat  $m_{odel}$ 5) arranged in series.

F, X<sub>0</sub>, S<sub>0</sub>



Consider an organism which follows the Monod equation where  $\mu_m = 0.5h^{-1}$  and  $K_s = 2 g/l$ 

a) In continuous perfectly mixed vessel at steady state with no cell death if  $S_0 = 50$  g/l and  $Y_{X/S} = 1$ (g cells/g substrate), what dilution rate D will give the maximum total rate of cell production?

- b) For the same value of D using tanks, find out cell concentration and substrate concentration (10)for first and second tank.
- In fed-batch fermentor, substrate stream is added continuously to the reactor. Develop a suitable 6. a) mathematical model with the following kinetics:

$$r_{X} = \mu X$$
,  $\mu = \frac{\mu_{\max}S}{K_{S} + S}$ ,  $r_{S} = -\frac{r_{X}}{Y}$ ,  $D = \frac{F_{0}}{V}$ 

Convert the model in the dimensionless using the following transformation:

$$V' = \frac{V}{V_0}, \quad X' = \frac{X}{YS_0}, \quad S' = \frac{S}{S_0}, \quad F' = \frac{F_0}{V_o\mu}, \quad K's = \frac{K_s}{S_0}, \quad \mu' = \frac{\mu}{\mu_m}, \quad F' = \frac{dV'}{dt'} \text{ and }$$

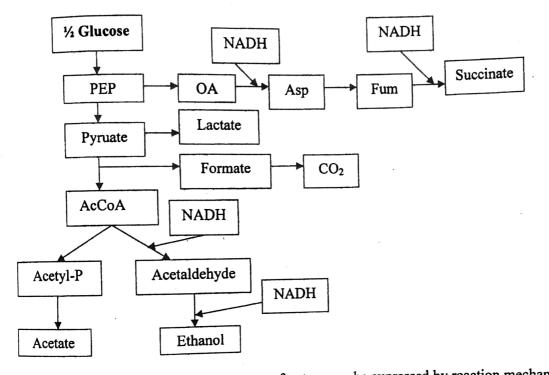
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(7, 3)

Discuss the Monod chemostat model with recycle system.

-3-

7. a) The metabolic pathway for aromatic amino acid synthesis is given bellow; identify the Substrate, products, intermediates. Construct the stoichiometric model and develop the matrix Form.



b)

Enzymatic isomerization glucose to fructose can be expressed by reaction mechanism:  $E + S \leftrightarrow ES \leftrightarrow E + P$ 

The kinetic parameter is:

 $\frac{V_{m,s}}{K_{m,s}} = 0.128$ ,  $\frac{V_{m,p}}{K_p} = 0.098$ ,  $\frac{1}{K_{m,s}} = 0.383$ ,  $\frac{1}{K_p} = 0.25$ 

If the feed (glucose) concentration is 1.0 kg mole/liter and desired conversion is 40%. Compare the productivity in above rate expression in CSTR & FPR. (4, 6)

b)